## 5.3 AIR QUALITY

This section discusses the regulatory setting, regional climate, topography, air pollution potential, and existing ambient air quality for criteria air pollutants, toxic air contaminants, odors, and dust. Information presented in this section is based in part on information gathered from the Bay Area Air Quality Management District (BAAQMD) and the California Air Resources Board (CARB).

## REGULATORY SETTING

Air quality with respect to criteria air pollutants and toxic air contaminants (TACs) within the San Francisco Bay Area Air Basin (SFBAAB) is regulated by such agencies as the BAAQMD, CARB, and Federal EPA. Each of these agencies develops rules, regulations, policies, and/or goals to attain the goals or directives imposed through legislation. Although the EPA regulations may not be superseded, both State and local regulations may be more stringent.

FEDERAL

## **U.S. Environmental Protection Agency**

At the Federal level, EPA has been charged with implementing national air quality programs. EPA's air quality mandates are drawn primarily from the Federal Clean Air Act (FCAA), which was enacted in 1963. The FCAA was amended in 1970, 1977, and 1990.

The FCAA required EPA to establish primary and secondary national ambient air quality standards (NAAQS). The FCAA also required each state to prepare an air quality control plan referred to as a State Implementation Plan (SIP). The Federal Clean Air Act Amendments of 1990 (FCAAA) added requirements for states with nonattainment areas to revise their SIPs to incorporate additional control measures to reduce air pollution. The SIP is periodically modified to reflect the latest emissions inventories, planning documents, and rules and regulations of the air basins as reported by their jurisdictional agencies. EPA has responsibility to review all state SIPs to determine conformity to the mandates of the FCAAA and determine if implementation will achieve air quality goals. If the EPA determines a SIP to be inadequate, a Federal Implementation Plan (FIP) may be prepared for the nonattainment area that imposes additional control measures. Failure to submit an approvable SIP or to implement the plan within the mandated timeframe may result in sanctions being applied to transportation funding and stationary air pollution sources in the air basin.

### Federal Hazardous Air Pollutant Program

Title III of the FCAA requires the EPA to promulgate national emissions standards for hazardous air pollutants (NESHAPs). The NESHAP may differ for major sources than for area sources of HAPs (major sources are defined as stationary sources with potential to emit more than 10 tons per year [TPY] of any HAP or more than 25 TPY of any combination of HAPs; all other sources are considered area sources). The emissions standards are to be promulgated in two phases. In the first phase (1992–2000), the EPA developed technology-based emission standards designed to produce the maximum emission reduction achievable. These standards are generally referred to as requiring maximum available control technology (MACT). These Federal rules are also commonly referred to as MACT standards, because they reflect the Maximum Achievable Control Technology. For area sources, the standards may be different, based on generally available control technology. In the second phase (2001–2008), the EPA is required to promulgate health risk—based emissions standards where deemed necessary to address risks remaining after implementation of the technology-based NESHAP standards. The FCAAA required the EPA to promulgate vehicle or fuel standards containing reasonable requirements that control toxic

emissions, at a minimum to benzene and formaldehyde. Performance criteria were established to limit mobile-source emissions of toxics, including benzene, formaldehyde, and 1,3-butadiene. In addition, §219 required the use of reformulated gasoline in selected U.S. cities (those with the most severe ozone nonattainment conditions) to further reduce mobile-source emissions.

### STATE

In 1992 and 1993, the California Air Resources Board (CARB) requested delegation of authority for the implementation and enforcement of specified New Source Performance Standards (NSPS) and National Emission Standards for Hazardous Air Pollutants (NESHAPS) to the following local agencies: Bay Area and South Coast Air Quality Management Districts (AQMDs). EPA's review of the State of California's laws, rules, and regulations showed them to be adequate for the implementation and enforcement of these Federal standards, and EPA granted the delegations as requested.

### California Air Resources Board

CARB is the agency responsible for coordination and oversight of State and local air pollution control programs in California and for implementing the California Clean Air Act (CCAA), which was adopted in 1988. The CCAA requires that all air districts in the State endeavor to achieve and maintain the CAAQS by the earliest practical date. The act specifies that districts should focus particular attention on reducing the emissions from transportation and area-wide emission sources, and provides districts with the authority to regulate indirect sources.

CARB is primarily responsible for developing and implementing air pollution control plans to achieve and maintain the NAAQS. CARB is primarily responsible for statewide pollution sources and produces a major part of the SIP. Local air districts are still relied upon to provide additional strategies for sources under their jurisdiction. The CARB combines this data and submits the completed SIP to EPA.

Other CARB duties include monitoring air quality (in conjunction with air monitoring networks maintained by air pollution control and air quality management districts), establishing CAAQS (which in many cases are more stringent than the NAAQS), determining and updating area designations and maps, and setting emissions standards for new mobile sources, consumer products, small utility engines, and off-road vehicles.

## **Transport of Pollutants**

The California Clean Air Act, Section 39610 (a), directs the CARB to "identify each district in which transported air pollutants from upwind areas outside the district cause or contribute to a violation of the ozone standard and to identify the district of origin of transported pollutants." The information regarding the transport of air pollutants from one basin to another was to be quantified to assist interrelated basins in the preparation of plans for the attainment of State ambient air quality standards. Numerous studies conducted by the CARB have identified air basins that are impacted by pollutants transported from other air basins (as of 1993). Among the air basins affected by air pollution transport from the SFBAAB are the North Central Coast Air Basin, the Mountain Counties Air Basin, the San Joaquin Valley Air Basin, and the Sacramento Valley Air Basin. The SFBAAB was also identified as an area impacted by the transport of air pollutants from the Sacramento region.

## **State Toxic Air Contaminant Programs**

California regulates TACs primarily through the Tanner Air Toxics Act (AB 1807) and the Air Toxics Hot Spots Information and Assessment Act of 1987 (AB 2588). The Tanner Act sets forth a formal procedure for CARB to designate substances as TACs. This includes research, public participation, and scientific

peer review before CARB can designate a substance as a TAC. To date, CARB has identified over 21 TACs, and adopted the EPA's list of HAPs as TACs. Most recently, diesel exhaust particulate was added to the CARB list of TACs. Once a TAC is identified, CARB then adopts an Airborne Toxics Control Measure for sources that emit that particular TAC. If there is a safe threshold for a substance at which there is no toxic effect, the control measure must reduce exposure below that threshold. If there is no safe threshold, the measure must incorporate best available control technology (BACT) to minimize emissions. None of the TACs identified by CARB have a safe threshold.

The Hot Spots Act requires that existing facilities that emit toxic substances above a specified level:

- 1. Prepare a toxic emission inventory;
- 2. Prepare a risk assessment if emissions are significant;
- 3. Notify the public of significant risk levels; and
- 4. Prepare and implement risk reduction measures.

CARB has adopted diesel exhaust control measures and more stringent emission standards for various on-road mobile sources of emissions, including transit buses and off-road diesel equipment (e.g., tractors and generators). In February 2000, CARB adopted a new public transit bus fleet rule and emission standards for new urban buses. These new rules and standards provide for: 1) more stringent emission standards for some new urban bus engines beginning with 2002 model year engines, 2) zeroemission bus demonstration and purchase requirements applicable to transit agencies, and 3) reporting requirements with which transit agencies must demonstrate compliance with the urban transit bus fleet rule. Upcoming milestones include the low sulfur diesel fuel requirement, and tighter emission standards for heavy-duty diesel trucks (2007) and off-road diesel equipment (2011) nationwide. Over time, the replacement of older vehicles will result in a vehicle fleet that produces substantially less TACs than under current conditions. Mobile-source emissions of TACs (e.g., benzene, 1-3-butadiene, and diesel PM) have been reduced significantly over the last decade, and will be reduced further in California through a progression of regulatory measures [e.g., Low Emission Vehicle/Clean Fuels and Phase II reformulated gasoline regulations) and control technologies. With implementation of CARB's Risk Reduction Plan, it is expected that diesel PM concentrations will be reduced by 75% in 2010 and 85% in 2020 from the estimated year 2000 level. Adopted regulations are also expected to continue to reduce formaldehyde emissions from cars and light-duty trucks. As emissions are reduced, it is expected that risks associated with exposure to the emissions will also be reduced.

**LOCAL** 

### **Bay Area Air Quality Management District**

The BAAQMD attains and maintains air quality conditions in the SFBAAB through a comprehensive program of planning, regulation, enforcement, technical innovation, and promotion of the understanding of air quality issues. The clean air strategy of the BAAQMD includes the preparation of plans for the attainment of ambient air quality standards, adoption and enforcement of rules and regulations concerning sources of air pollution, and issuance of permits for stationary sources of air pollution. The BAAQMD also inspects stationary sources of air pollution and responds to citizen complaints, monitors ambient air quality and meteorological conditions, and implements programs and regulations required by the FCAA, FCAAA, and the CCAA.

The BAAQMD has regulated TACs since the 1980s. At the local level, air pollution control or management districts may adopt and enforce CARB's control measures. Under BAAQMD Regulation 2-1

(General Permit Requirements), Regulation 2-2 (New Source Review), and Regulation 2-5 (New Source Review), all nonexempt sources that possess the potential to emit TACs are required to obtain permits from BAAQMD. Permits may be granted to these operations if they are constructed and operated in accordance with applicable regulations, including new source review standards and air toxics control measures. The BAAQMD limits emissions and public exposure to TACs through a number of programs. The BAAQMD prioritizes TAC-emitting stationary sources based on the quantity and toxicity of the TAC emissions and the proximity of the facilities to sensitive receptors. In addition, the BAAQMD has adopted Regulation 11 Rules 2 and 14, which address asbestos demolition renovation, manufacturing, and standards for asbestos containing serpentine.

### **BAAQMD Air Quality Plans**

As stated above, the BAAQMD prepares plans to attain ambient air quality standards in the SFBAAB. The BAAQMD prepares ozone attainment plans (OAP) for the national ozone standard and clean air plans (CAP) for the California standard both in coordination with the Metropolitan Transportation Commission and the Association of Bay Area Governments (ABAG).

With respect to applicable air quality plans, the BAAQMD prepared the 2010 Clean Air Plan to address nonattainment of the national 1-hour ozone standard in the SFBAAB. The purpose of the 2010 Clean Air Plan is to:

- 1. Update the Bay Area 2005 Ozone Strategy in accordance with the requirements of the California Clean Air Act to implement "all feasible measures" to reduce ozone;
- 2. Consider the impacts of ozone control measures on particulate matter (PM), air toxics, and greenhouse gases in a single, integrated plan;
- 3. Review progress in improving air quality in recent years;
- 4. Establish emission control measures to be adopted or implemented in the 2009-2012 timeframe; and
- 5. Address nonattainment of the CAAQS.

### **City of Milpitas General Plan**

The existing City of Milpitas General Plan identifies the following policies related to air quality:

### **Housing Element**

- GOAL F-1: Promote Energy Conservation in Residential Development. The City of Milpitas will promote energy efficiency in residential development within the City, including reduction of energy use through better design and construction in individual homes, and also through energy efficient urban design.
  - **Policy F-1:** The City will continue to undertake a variety of activities to achieve energy efficiency in residential development in conformance with State laws.

The City will continue to partner with local utility providers to promote participation of Milpitas' low-income residents in available energy efficiency programs, such as PG&E's Energy Partners Program.

The City will continue to promote use of passive solar devices and promote energy audits of existing homes.

The City will adopt a Green Building Ordinance by the end of 2009.

The City will continue to encourage the incorporation of energy-saving principles in the design and planning of new residential developments, including features such as solar orientation.

The City will continue to encourage mixed-use and transit-oriented development at transit nodes.

In accordance with the Green Building Policy Resolution adopted in February 2008, the City will continue to require that planning applications for new buildings include a completed LEED or GreenPoint Rated checklist.

### ENVIRONMENTAL SETTING

### San Francisco Bay Area Air Basin

Milpitas is located within the San Francisco Bay Area Air Basin (SFBAAB), which comprises all of Alameda, Contra Costa, Marin, Napa, San Francisco, San Mateo, and Santa Clara counties, the southern portion of Sonoma County, and the southwestern portion of Solano County. Air quality in this area is determined by such natural factors as topography, meteorology, and climate, in addition to the presence of existing air pollution sources and ambient conditions. These factors along with applicable regulations are discussed below.

CLIMATE, TOPOGRAPHY, AND AIR POLLUTION POTENTIAL

The SFBAAB is characterized by complex terrain, consisting of coastal mountain ranges, inland valleys, and bays, which distort normal wind flow patterns.

The climate is dominated by the strength and location of a semi-permanent, subtropical high-pressure cell. During the summer, the Pacific high pressure cell is centered over the northeastern portion of the Pacific Ocean, resulting in stable meteorological conditions and a steady northwesterly wind flow. Upwelling of cold ocean water from below to the surface because of the northwesterly flow produces a band of cold water off the California coast. The cool and moisture-laden air approaching the coast from the Pacific Ocean is further cooled by the presence of the cold water band resulting in condensation and the presence of fog and stratus clouds along the Northern California coast.

In the winter, the Pacific high-pressure cell weakens and shifts southward resulting in wind flow offshore, the absence of upwelling, and the occurrence of storms. Weak inversions coupled with moderate winds result in a low air pollution potential.

### High Pressure Cell

During the summer, the large-scale meteorological condition that dominates the West Coast is a semipermanent high pressure cell centered over the northeastern portion of the Pacific Ocean. This high pressure cell keeps storms from affecting the California coast. Hence, the SFBAAB experiences little precipitation in the summer months. Winds tend to blow on shore out of the north/northwest.

The steady northwesterly flow induces upwelling of cold water from below. This upwelling produces a band of cold water off the California coast. When air approaches the California coast, already cool and moisture-laden from its long journey over the Pacific, it is further cooled as it crosses this bank of cold water. This cooling often produces condensation resulting in a high incidence of fog and stratus clouds along the Northern California coast in the summer.

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Generally in the winter, the Pacific high pressure cell weakens and shifts southward, winds tend to flow offshore, upwelling ceases, and storms occur. During the winter rainy periods, inversions (layers of warmer air over colder air; see below) are weak or nonexistent, winds are usually moderate, and air pollution potential is low. The Pacific high pressure cell does periodically become dominant, bringing strong inversions, light winds, and high pollution potential.

### **Topography**

The topography of the SFBAAB is characterized by complex terrain, consisting of coastal mountain ranges, inland valleys, and bays. This complex terrain, especially the higher elevations, distorts the normal wind flow patterns in the SFBAAB. The greatest distortion occurs when low-level inversions are present and the air beneath the inversion flows independently of air above the inversion, a condition that is common in the summer time.

The only major break in California's Coast Range occurs in the SFBAAB. Here the Coast Range splits into western and eastern ranges. Between the two ranges lies San Francisco Bay. The gap in the western coast range is known as the Golden Gate, and the gap in the eastern coast range is the Carquinez Strait. These gaps allow air to pass into and out of the SFBAAB and the Central Valley.

### **Wind Patterns**

During the summer, winds flowing from the northwest are drawn inland through the Golden Gate and over the lower portions of the San Francisco Peninsula. Immediately south of Mount Tamalpais, the northwesterly winds accelerate considerably and come more directly from the west as they stream through the Golden Gate. This channeling of wind through the Golden Gate produces a jet that sweeps eastward and splits off to the northwest toward Richmond and to the southwest toward San Jose when it meets the East Bay hills.

Wind speeds may be strong locally in areas where air is channeled through a narrow opening, such as the Carquinez Strait, the Golden Gate, or the San Bruno gap. For example, the average wind speed at San Francisco International Airport in July is about 17 knots (from 3 p.m. to 4 p.m.), compared with only 7 knots at San Jose and less than 6 knots at the Farallon Islands.

The air flowing in from the coast to the Central Valley, called the sea breeze, begins developing at or near ground level along the coast in late morning or early afternoon. As the day progresses, the sea breeze layer deepens and increases in velocity while spreading inland. The depth of the sea breeze depends in large part upon the height and strength of the inversion. If the inversion is low and strong, and hence stable, the flow of the sea breeze will be inhibited and stagnant conditions are likely to result.

In the winter, the SFBAAB frequently experiences stormy conditions with moderate to strong winds, as well as periods of stagnation with very light winds. Winter stagnation episodes are characterized by nighttime drainage flows in coastal valleys. Drainage is a reversal of the usual daytime air-flow patterns; air moves from the Central Valley toward the coast and back down toward the Bay from the smaller valleys within the SFBAAB.

### **Temperature**

Summertime temperatures in the SFBAAB are determined in large part by the effect of differential heating between land and water surfaces. Because land tends to heat up and cool off more quickly than water, a large-scale gradient (differential) in temperature is often created between the coast and the Central Valley, and small-scale local gradients are often produced along the shorelines of the ocean and bays. The temperature gradient near the ocean is also exaggerated, especially in summer, because of the upwelling of cold ocean bottom water along the coast. On summer afternoons the temperatures at

the coast can be 35°F cooler than temperatures 15 to 20 miles inland. At night this contrast usually decreases to less than 10°.

In the winter, the relationship of minimum and maximum temperatures is reversed. During the daytime the temperature contrast between the coast and inland areas is small, whereas at night the variation in temperature is large.

### **Precipitation**

The SFBAAB is characterized by moderately wet winters and dry summers. Winter rains account for about 75 percent of the average annual rainfall. The amount of annual precipitation can vary greatly from one part of the SFBAAB to another even within short distances. In general, total annual rainfall can reach 40 inches in the mountains, but it is often less than 16 inches in sheltered valleys.

During rainy periods, ventilation (rapid horizontal movement of air and injection of cleaner air) and vertical mixing are usually high, and thus pollution levels tend to be low. However, frequent dry periods do occur during the winter where mixing and ventilation are low and pollutant levels build up.

### **Air Pollution Potential**

The potential for high pollutant concentrations developing at a given location depends upon the quantity of pollutants emitted into the atmosphere in the surrounding area or upwind, and the ability of the atmosphere to disperse the contaminated air. The topographic and climatological factors discussed above influence the atmospheric pollution potential of an area. Atmospheric pollution potential, as the term is used here, is independent of the location of emission sources and is instead a function of factors described below.

#### Wind Circulation

Low wind speed contributes to the buildup of air pollution because it allows more pollutants to be emitted into the air mass per unit of time. Light winds occur most frequently during periods of low sun (fall and winter, and early morning) and at night. These are also periods when air pollutant emissions from some sources are at their peak, namely, commute traffic (early morning) and wood burning appliances (nighttime). The problem can be compounded in valleys, when weak flows carry the pollutants upvalley during the day, and cold air drainage flows move the air mass downvalley at night. Such restricted movement of trapped air provides little opportunity for ventilation and leads to buildup of pollutants to potentially unhealthful levels.

### Inversions

An inversion is a layer of warmer air over a layer of cooler air. Inversions affect air quality conditions significantly because they influence the mixing depth (i.e., the vertical depth in the atmosphere available for diluting air contaminants near the ground). The highest air pollutant concentrations in the SFBAAB generally occur during inversions.

There are two types of inversions that occur regularly in the SFBAAB. One is more common in the summer and fall, while the other is most common during the winter. The frequent occurrence of elevated temperature inversions in summer and fall months acts to cap the mixing depth, limiting the depth of air available for dilution. Elevated inversions are caused by subsiding air from the subtropical high pressure zone, and from the cool marine air layer that is drawn into the SFBAAB by the heated low pressure region in the Central Valley.

The inversions typical of winter, called radiation inversions, are formed as heat quickly radiates from the earth's surface after sunset, causing the air in contact with it to rapidly cool. Radiation inversions are

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strongest on clear, low-wind, cold winter nights, allowing the build-up of such pollutants as carbon monoxide and particulate matter. When wind speeds are low, there is little mechanical turbulence to mix the air, resulting in a layer of warm air over a layer of cooler air next to the ground. Mixing depths under these conditions can be as shallow as 50 to 100 meters, particularly in rural areas. Urban areas usually have deeper minimum mixing layers because of heat island effects and increased surface roughness. During radiation inversions downwind transport is slow, the mixing depths are shallow, and turbulence is minimal, all factors which contribute to ozone formation.

Although each type of inversion is most common during a specific season, either inversion mechanism can occur at any time of the year. Sometimes both occur simultaneously. Moreover, the characteristics of an inversion often change throughout the course of a day. The terrain of the SFBAAB also induces significant variations among subregions.

### Solar Radiation

The frequency of hot, sunny days during the summer months in the SFBAAB is another important factor that affects air pollution potential. It is at the higher temperatures that ozone is formed. In the presence of ultraviolet sunlight and warm temperatures, reactive organic gases and oxides of nitrogen react to form secondary photochemical pollutants, including ozone. Because temperatures in many of the SFBAAB inland valleys are so much higher than near the coast, the inland areas are especially prone to photochemical air pollution.

In late fall and winter, solar angles are low, resulting in insufficient ultraviolet light and warming of the atmosphere to drive the photochemical reactions. Ozone concentrations do not reach significant levels in the SFBAAB during these seasons.

### Sheltered Terrain

The hills and mountains in the SFBAAB contribute to the high pollution potential of some areas. During the day, or at night during windy conditions, areas in the lee sides of mountains are sheltered from the prevailing winds, thereby reducing turbulence and downwind transport. At night, when wind speeds are low, the upper atmospheric layers are often decoupled from the surface layers during radiation conditions. If elevated terrain is present, it will tend to block pollutant transport in that direction. Elevated terrain also can create a recirculation pattern by inducing upvalley air flows during the day and reverse downvalley flows during the night, allowing little inflow of fresh air.

The areas having the highest air pollution potential tend to be those that experience the highest temperatures in the summer and the lowest temperatures in the winter. The coastal areas are exposed to the prevailing marine air, creating cooler temperatures in the summer, warmer temperatures in winter, and stratus clouds all year. The inland valleys are sheltered from the marine air and experience hotter summers and colder winters. Thus, the topography of the inland valleys creates conditions conducive to high air pollution potential.

### Pollution Potential Related to Emissions

Although air pollution potential is strongly influenced by climate and topography, the air pollution that occurs in a location also depends upon the amount of air pollutant emissions in the surrounding area or transported from more distant places. Air pollutant emissions generally are highest in areas that have high population densities, high motor vehicle use, and/or industrialization. These contaminants created by photochemical processes in the atmosphere, such as ozone, may result in high concentrations many miles downwind from the sources of their precursor chemicals.

### Santa Clara Valley Climatological Subregion

There are 11 climatological subregions within the SFBAAB. Milpitas is located within the Santa Clara Valley subregion. The Santa Clara Valley is bounded by the Bay to the north and by mountains to the east, south and west. Temperatures are warm on summer days and cool on summer nights, and winter temperatures are fairly mild. At the northern end of the valley, mean maximum temperatures are in the low-80's during the summer and the high-50's during the winter, and mean minimum temperatures range from the high-50's in the summer to the low-40's in the winter. Further inland, where the moderating effect of the Bay is not as strong, temperature extremes are greater. For example, in San Martin, located 27 miles south of the San Jose Airport, temperatures can be more than 10 degrees warmer on summer afternoons and more than 10 degrees cooler on winter nights.

Winds in the valley are greatly influenced by the terrain, resulting in a prevailing flow that roughly parallels the valley's northwest-southeast axis. A north-northwesterly sea breeze flows through the valley during the afternoon and early evening, and a light south-southeasterly drainage flow occurs during the late evening and early morning. In the summer the southern end of the valley sometimes becomes a "convergence zone," when air flowing from the Monterey Bay gets channeled northward into the southern end of the valley and meets with the prevailing northwesterly winds.

Wind speeds are greatest in the spring and summer and weakest in the fall and winter. Nighttime and early morning hours frequently have calm winds in all seasons, while summer afternoons and evenings are quite breezy. Strong winds are rare, associated mostly with the occasional winter storm.

The air pollution potential of the Santa Clara Valley is high. High summer temperatures, stable air and mountains surrounding the valley combine to promote ozone formation. In addition to the many local sources of pollution, ozone precursors from San Francisco, San Mateo and Alameda Counties are carried by prevailing winds to the Santa Clara Valley. The valley tends to channel pollutants to the southeast. In addition, on summer days with low level inversions, ozone can be recirculated by southerly drainage flows in the late evening and early morning and by the prevailing northwesterlies in the afternoon. A similar recirculation pattern occurs in the winter, affecting levels of carbon monoxide and particulate matter. This movement of the air up and down the valley increases the impact of the pollutants significantly.

Pollution sources are plentiful and complex in this subregion. The Santa Clara Valley has a high concentration of industry at the northern end, in the Silicon Valley. Some of these industries are sources of air toxics as well as criteria air pollutants. In addition, Santa Clara Valley's large population and many work-site destinations generate the highest mobile source emissions of any subregion in the SFBAAB.

# EXISTING AMBIENT AIR QUALITY: CRITERIA AIR POLLUTANTS

The California Air Resources Board (CARB) and the U.S. Environmental Protection Agency (EPA) currently focus on the following air pollutants as indicators of ambient air quality: ozone  $(O_3)$ , particulate matter (PM), nitrogen dioxide  $(NO_2)$ , carbon monoxide (CO), sulfur dioxide  $(SO_2)$ , and lead (Pb). Because these are the most prevalent air pollutants known to be deleterious to human health, they are commonly referred to as "criteria air pollutants." Sources and health effects of the criteria air pollutants are summarized in Table 5.3-1.

TABLE 5.3-1: COMMON SOURCES OF HEALTH EFFECTS FOR CRITERIA AIR POLLUTANTS

| POLLUTANTS  | Sources   | HEALTH EFFECTS   |
|---|---|--|
| Ozone (O₃)  | Atmospheric reaction of organic gases with nitrogen oxides in sunlight  | Aggravation of respiratory and cardiovascular diseases; reduced lung function; increased cough and chest discomfort                                  |
| Fine Particulate Matter (PM <sub>10</sub> and PM <sub>2.5</sub> ) | Stationary combustion of solid<br>fuels; construction activities;<br>industrial processes;<br>atmospheric chemical reactions                                  | Reduced lung function; aggravation of respiratory and cardiovascular diseases; increases in mortality rate; reduced lung function growth in children |
| Nitrogen Dioxide (NO <sub>2</sub> )                               | Motor vehicle exhaust; high temperature stationary combustion; atmospheric reactions  | Aggravation of respiratory illness   |
| Carbon Monoxide (CO)  | Incomplete combustion of fuels and other carbon-containing substances, such as motor vehicle exhaust; natural events, such as decomposition of organic matter | Aggravation of some heart diseases; reduced tolerance for exercise; impairment of mental function; birth defects; death at high levels of exposure   |
| Sulfur Dioxide (SO <sub>2</sub> )                                 | Combination of sulfur-<br>containing fossil fuels; smelting<br>of sulfur-bearing metal ore;<br>industrial processes   | Aggravation of respiratory diseases; reduced lung function   |
| Lead (Pb)   | Contaminated soil   | Behavioral and hearing disabilities in children; nervous system impairment   |

Source: Bay Area Air Quality Management District (2012)

**Ozone** ( $O_3$ ), or smog, is not emitted directly into the environment, but is formed in the atmosphere by complex chemical reactions between reactive organic gases (ROG) and nitrous oxide (NO<sub>x</sub>) in the presence of sunlight. Ozone formation is greatest on warm, windless, sunny days. The main sources of NO<sub>x</sub> and ROG, often referred to as ozone precursors, are combustion processes (including motor vehicle engines), the evaporation of solvents, paints, and fuels, and biogenic sources. Automobiles are the single largest source of ozone precursors in the SFBAAB. Tailpipe emissions of ROG are highest during cold starts, hard acceleration, stop-and-go conditions, and slow speeds. They decline as speeds increase up to about 50 mph, then increase again at high speeds and high engine loads. ROG emissions associated with evaporation of unburned fuel depend on vehicle and ambient temperature cycles. Nitrogen oxide emissions exhibit a different curve; emissions decrease as the vehicle approaches 30 mph and then begin to increase with increasing speeds.

Ozone levels usually build up during the day and peak in the afternoon hours. Short-term exposure can irritate the eyes and cause constriction of the airways. Besides causing shortness of breath, it can aggravate existing respiratory diseases such as asthma, bronchitis, and emphysema. Chronic exposure to high ozone levels can permanently damage lung tissue. Ozone can also damage plants and trees, and materials such as rubber and fabrics.

**Particulate Matter (PM)** refers to a wide range of solid or liquid particles in the atmosphere, including smoke, dust, aerosols, and metallic oxides. Respirable particulate matter with an aerodynamic diameter of 10 micrometers or less is referred to as  $PM_{10}$ .  $PM_{2.5}$  includes a subgroup of finer particles that have an aerodynamic diameter of 2.5 micrometers or less. Some particulate matter, such as pollen, is naturally

occurring. In the SFBAAB most particulate matter is caused by combustion, factories, construction, grading, demolition, agricultural activities, and motor vehicles. Extended exposure to particulate matter can increase the risk of chronic respiratory disease.  $PM_{10}$  is of concern because it bypasses the body's natural filtration system more easily than larger particles, and can lodge deep in the lungs. The EPA and the State of California revised their PM standards several years ago to apply only to these fine particles.  $PM_{2.5}$  poses an increased health risk because the particles can deposit deep in the lungs and contain substances that are particularly harmful to human health. Motor vehicles are currently responsible for about half of particulates in the SFBAAB. Wood burning in fireplaces and stoves is another large source of fine particulates.

**Nitrogen Dioxide** (NO<sub>2</sub>) is a reddish-brown gas that is a by-product of combustion processes. Automobiles and industrial operations are the main sources of NO<sub>2</sub>. Aside from its contribution to ozone formation, nitrogen dioxide can increase the risk of acute and chronic respiratory disease and reduce visibility. NO<sub>2</sub> may be visible as a coloring component of a brown cloud on high pollution days, especially in conjunction with high ozone levels.

Carbon Monoxide (CO) is an odorless, colorless gas. It is formed by the incomplete combustion of fuels. The single largest source of CO in the SFBAAB is motor vehicles. Emissions are highest during cold starts, hard acceleration, stop-and-go driving, and when a vehicle is moving at low speeds. New findings indicate that CO emissions per mile are lowest at about 45 mph for the average light-duty motor vehicle and begin to increase again at higher speeds. When inhaled at high concentrations, CO combines with hemoglobin in the blood and reduces the oxygen-carrying capacity of the blood. This results in reduced oxygen reaching the brain, heart, and other body tissues. This condition is especially critical for people with cardiovascular diseases, chronic lung disease or anemia, as well as fetuses. Even healthy people exposed to high CO concentrations can experience headaches, dizziness, fatigue, unconsciousness, and even death.

**Sulfur Dioxide (SO<sub>2</sub>)** is a colorless acid gas with a pungent odor. It has potential to damage materials and it can have health effects at high concentrations. It is produced by the combustion of sulfur-containing fuels, such as oil, coal, and diesel. SO<sub>2</sub> can irritate lung tissue and increase the risk of acute and chronic respiratory disease.

**Lead (Pb)** is a metal found naturally in the environment as well as in manufactured products. The major sources of lead emissions have historically been mobile and industrial sources. As a result of the phase-out of leaded gasoline, metal processing is currently the primary source of lead emissions. The highest levels of lead in air are generally found near lead smelters. Other stationary sources are waste incinerators, utilities, and lead-acid battery manufacturers.

Twenty years ago, mobile sources were the main contributor to ambient lead concentrations in the air. In the early 1970s, the EPA set national regulations to gradually reduce the lead content in gasoline. In 1975, unleaded gasoline was introduced for motor vehicles equipped with catalytic converters. The EPA banned the use of leaded gasoline in highway vehicles in December 1995. As a result of the EPA's regulatory efforts to remove lead from gasoline, emissions of lead from the transportation sector and levels of lead in the air decreased dramatically.

### **Ambient Air Quality Standards and Designations**

The current Federal and State ambient air quality standards and attainment standards are presented in Table 5.3-2.

TABLE 5.3-2: AMBIENT AIR QUALITY STANDARDS AND DESIGNATIONS

|   |                              | CALIFORNIA  | NATIONAL STANDARDS A |                             |                                |                                   |
|---|------------------------------|---|----------------------|-----------------------------|--------------------------------|-----------------------------------|
| POLLUTANT                                     | AVERAGING TIME               | STANDARDS <sup>B, C</sup>   | ATTAINMENT STATUS D  | PRIMARY <sup>C,E</sup>      | SECONDARY C,F                  | ATTAINMENT<br>STATUS <sup>G</sup> |
|   | 1-hour                       | 0.09 ppm (180 μg/m³)  | N (Serious)          | – h                         | Same as                        | – h                               |
| Ozone   | 8-hour                       | 0.070 ppm (137 μg/m³)   | _                    | 0.075 ppm<br>(147<br>μg/m³) | Primary<br>Standard            | N                                 |
| Carbon<br>Monoxide                            | 1-hour                       | our 20 ppm (23 mg/m³) A   |                      | 35 ppm<br>(40<br>mg/m³)     | _                              | U/A                               |
| (CO)  | 8-hour                       | 9 ppm (10 mg/m³)  |                      | 9 ppm (10<br>mg/m³)         |                                |                                   |
| Nitrogen<br>Dioxide                           | Annual<br>Arithmetic<br>Mean | 0.030 ppm (57 μg/m³)  | -                    | 0.053 ppm<br>(100<br>μg/m³) | Same as<br>Primary<br>Standard | U/A                               |
| (NO <sub>2</sub> )                            | 1-hour                       | 0.18 ppm (339 μg/m³)  | Α                    | _                           | Standard                       | _                                 |
|   | Annual<br>Arithmetic<br>Mean | -   | -                    | 0.030 ppm<br>(80 μg/m³)     | _                              |                                   |
| Sulfur<br>Dioxide                             | 24-hour                      | 0.04 ppm (105 μg/m³)  | А                    | 0.14 ppm<br>(365<br>μg/m³)  | _                              | А                                 |
| (SO <sub>2</sub> )                            | 3-hour                       | _   | _                    | _                           | 0.5 ppm<br>(1300<br>μg/m³)     |                                   |
|   | 1-hour                       | 0.25 ppm (655 μg/m <sup>3</sup> )   | Α                    | _                           | _                              | _                                 |
| Respirable<br>Particulate<br>Matter           | Annual<br>Arithmetic<br>Mean | 20 μg/m³  | N                    | - h                         | Same as<br>Primary             | U                                 |
| (PM <sub>10</sub> )                           | 24-hour                      | 50 μg/m <sup>3</sup>  |                      | 150 μg/m <sup>3</sup>       | g/m <sup>3</sup> Standard      |                                   |
| Fine<br>Particulate<br>Matter                 | Annual<br>Arithmetic<br>Mean | 12 μg/m³  | N                    | 15 μg/m³                    | Same as<br>Primary             | N                                 |
| (PM <sub>2.5</sub> )                          | 24-hour                      | _   | _                    | 35 μg/m <sup>3</sup>        | Standard                       |                                   |
|   | 30-day<br>Average            | 1.5 μg/m³   | А                    | _                           | _                              | _                                 |
| Lead  | Calendar<br>Quarter          | -   | -                    | 1.5 μg/ m <sup>3</sup>      | Same as<br>Primary<br>Standard | _                                 |
| Sulfates                                      | 24-hour                      | 25 μg/m³  | Α                    |                             |                                |                                   |
| Hydrogen<br>Sulfide                           | 1-hour                       | 0.03 ppm (42 μg/m³)   | U                    | No<br>National              |                                |                                   |
| Vinyl<br>Chloride                             | 24-hour                      | 0.01 ppm (26 μg/m³)   | -                    | Standards                   |                                |                                   |
| Visibility-<br>Reducing<br>Particle<br>Matter | 8-hour                       | Extinction coefficient of 0.23 per kilometer — visibility of 10 miles or more (0.07—30 miles or more for Lake Tahoe) because of particles | U                    |                             |                                |                                   |

|           | CALIFORNIA     |                           |                                   | N.                     | ATIONAL STANDARDS        | ; A                               |
|-----------|----------------|---------------------------|-----------------------------------|------------------------|--------------------------|-----------------------------------|
| POLLUTANT | AVERAGING TIME | STANDARDS <sup>B, C</sup> | ATTAINMENT<br>STATUS <sup>D</sup> | PRIMARY <sup>C,E</sup> | SECONDARY <sup>C,F</sup> | ATTAINMENT<br>STATUS <sup>G</sup> |
|           |                | when the relative         |                                   |                        |                          |                                   |
|           |                | humidity is less than     |                                   |                        |                          |                                   |
|           |                | 70%.                      |                                   |                        |                          |                                   |

- a National standards (other than ozone, PM, and those based on annual averages or annual arithmetic means) are not to be exceeded more than once a year. The ozone standard is attained when the fourth highest 8-hour concentration in a year, averaged over 3 years, is equal to or less than the standard. The PM<sub>10</sub> 24-hour standard is attained when 99% of the daily concentrations, averaged over 3 years, are equal to or less than the standard. The PM<sub>2.5</sub> 24-hour standard is attained when 98% of the daily concentrations, averaged over 3 years, are equal to or less than the standard. Contact the EPA for further clarification and current Federal policies.
- b California standards for ozone, CO (except Lake Tahoe), SO<sub>2</sub> (1- and 24-hour), NO<sub>2</sub>, PM, and visibility-reducing particles are values that are not to be exceeded. All others are not to be equaled or exceeded. CAAQS are listed in the Table of Standards in Section 70200 of Title 17 of the California Code of Regulations.
- c Concentration expressed first in units in which it was promulgated [i.e., parts per million (ppm) or micrograms per cubic meter (µg/m³)]. Equivalent units given in parentheses are based upon a reference temperature of 25°C and a reference pressure of 760 torr. Most measurements of air quality are to be corrected to a reference temperature of 25°C and a reference pressure of 760 torr; ppm in this table refers to ppm by volume, or micromoles of pollutant per mole of gas.
- d Unclassified (U): a pollutant is designated unclassified if the data are incomplete and do not support a designation of attainment or nonattainment.

Attainment (A): a pollutant is designated attainment if the State standard for that pollutant was not violated at any site in the area during a 3-year period.

Nonattainment (N): a pollutant is designated nonattainment if there was a least one violation of a State standard for that pollutant in the area.

Nonattainment/Transitional (NT): is a subcategory of the nonattainment designation. An area is designated nonattainment/transitional to signify that the area is close to attaining the standard for that pollutant.

- e National Primary Standards: The levels of air quality necessary, with an adequate margin of safety, to protect the public health.
- f National Secondary Standards: The levels of air quality necessary to protect the public welfare from any known or anticipated adverse effects of a pollutant.
- g Nonattainment (N): any area that does not meet (or that contributes to ambient air quality in a nearby area that does not meet) the national primary or secondary ambient air quality standard for the pollutant.

Attainment (A): any area that meets the national primary or secondary ambient air quality standard for the pollutant.

Unclassifiable (U): any area that cannot be classified on the basis of available information as meeting or not meeting the national primary or secondary ambient air quality standard for the pollutant.

- h The 1-hour ozone NAAQS was revoked on June 15, 2005 and the annual PM $_{10}$  NAAQS was revoked in 2006.
- i ARB has identified lead and vinyl chloride as toxic air contaminants with no threshold of exposure for adverse health effects determined. These actions allow for the implementation of control measures at levels below the ambient concentrations specified for this pollutant.
- j U.S EPA lowered the 24-hour PM<sub>2.5</sub> standard from 65  $\mu$ g/m³ to 35  $\mu$ g/m³ in 2006. EPA issued attainment status designations for the 35  $\mu$ g/m³ standard on December 22, 2008. EPA has designated the Bay Area as nonattainment for the 35  $\mu$ g/m³ PM<sub>2.5</sub> standard. The EPA designation will be effective 90 days after publication of the regulation in the Federal Register.

Source: Bay Area Air Quality Management District (2012)

### **Monitoring Data**

The BAAQMD operates a regional air quality monitoring network that regularly measures the concentrations of the major criteria air pollutants. Air pollutant monitoring data is available at http://www.arb.ca.gov/adam/welcome.html. Air quality conditions in the SFBAAB have improved significantly since the BAAQMD was created in 1955. Ambient concentrations and the number of days on which the region exceeds standards have declined dramatically. Neither Federal or State ambient air quality standards have been violated in recent decades for nitrogen dioxide, sulfur dioxide, sulfates, lead, hydrogen sulfide, and vinyl chloride.

The closest air quality monitoring site to Milpitas is located in San Jose – Jackson Street, approximately 5.6 miles to the south. This site measures ozone, PM<sub>10</sub>, and PM<sub>2.5</sub>. The next closest monitoring site is located in San Jose – Knox Avenue. This site measures PM<sub>2.5</sub> only. It is important to note that the Federal ozone 1-hour standard was revoked by the EPA and is no longer applicable for Federal standards. Data obtained from the monitoring sites between 2013 and 2015 is shown in Tables 5.3-3 and 5.3-4.

TABLE 5.3-3: AMBIENT AIR QUALITY MONITORING DATA (SAN JOSE - JACKSON STREET)

| THE DIE OIL OF THE DIEN     | TITIN QUILLITT PIONIT             | Omna Dilli (Billi Jobi             | Jiichbon bindaij |                    |
|-----------------------------|-----------------------------------|------------------------------------|------------------|--------------------|
| POLLUTANT                   | CAL.                              | FED. YEAR                          |                  | DAYS EXCEEDED      |
| 1 0220171141                | PRIMARY STANDARD                  |                                    | 7 27 (1)         | STATE/FED STANDARD |
| Ozone (O₃)                  |                                   |                                    | 2015             | 0/0                |
| (1-hour)                    | 0.09 ppm for 1 hour               | N/A                                | 2014             | 0/0                |
| (1-11001)                   |                                   |                                    | 2013             | 0/0                |
| Ozone (O <sub>3</sub> )     |                                   |                                    | 2015             | 2/2                |
| (8-hour)                    | 0.07 ppm for 8 hour               | 0.075 ppm for 8 hour               | 2014             | 0/0                |
| (8-11001)                   | (8-11041)                         |                                    | 2013             | 1/1                |
| Particulate                 |                                   |                                    | 2015             | * / *              |
| Matter (PM <sub>10</sub> )  | 50 ug/m <sup>3</sup> for 24 hours | 150 ug/m <sup>3</sup> for 24 hours | 2014             | 3.1 / 0            |
| Matter (PM10)               |                                   |                                    | 2013             | 15.2 / 0           |
| Particulate                 | No 24 hour State                  |                                    | 2015             | N/A / 2.1          |
| Matter (PM <sub>2.5</sub> ) | No 24 nour State Standard         | 35 ug/m³ for 24 hours              | 2014             | N/A / 2.0          |
| iviatiei (PIVI2.5)          | Standard                          |                                    | 2013             | N/A / 4.0          |

Sources: California Air Resources Board (ADAM) Air Pollution Summaries, 2013-2015.

TABLE 5.3-4: AMBIENT AIR QUALITY MONITORING DATA (SAN JOSE – KNOX AVENUE)

| POLLUTANT                                  | CAL.                              | FED.                               | YEAR                        | Days Exceeded                     |
|--|-----------------------------------|------------------------------------|-----------------------------|-----------------------------------|
| FOLLUTAINT                                 | Primary.                          | STANDARD                           | TEAR                        | STATE/FED STANDARD                |
| Ozone (O₃)<br>(1-hour)                     | 0.09 ppm for 1 hour               | N/A Not collected at t             |                             | d at this site.                   |
| Ozone (O₃)<br>(8-hour)                     | 0.07 ppm for 8 hour               | 0.075 ppm for 8 hour               | Not collected at this site. |                                   |
| Particulate<br>Matter (PM <sub>10</sub> )  | 50 ug/m <sup>3</sup> for 24 hours | 150 ug/m <sup>3</sup> for 24 hours | Not collecte                | d at this site.                   |
| Particulate<br>Matter (PM <sub>2.5</sub> ) | No 24 hour State<br>Standard      | 35 ug/m³ for 24 hours              | 2015<br>2014<br>2013        | N/A / 1.0<br>N/A / *<br>N/A / N/A |

Sources: California Air Resources Board (ADAM) Air Pollution Summaries, 2013-2015.

Notes:

PPM = PARTS PER MILLION.

 $UG/M^3 = MICRONS PER CUBIC METER.$ 

*N/A= NOT APPLICABLE* 

## **Emissions Inventory**

The BAAQMD estimates emissions of criteria air pollutants from approximately 900 source categories. The estimates are based on BAAQMD permit information for stationary sources (e.g., manufacturing industries, refineries, and dry-cleaning operations), plus more generalized estimates for area sources

<sup>\* =</sup> There was insufficient (or no) data available to determine the value

(e.g., space heating, landscaping activities, and use of consumer products) and mobile sources (e.g., trains, ships and planes, as well as on-road and off-road motor vehicles). BAAQMD emissions inventory data is available at http://www.arb.ca.gov/app/emsinv/emssumcat.php. Table 5.3-5 presents the 2015 estimated annual emissions in Santa Clara County.

TABLE 5.3-5: 2012 ESTIMATED ANNUAL AVERAGE EMISSIONS (SANTA CLARA COUNTY)

| STATIONARY SOURCES                    | TOG    | ROG   | со     | $NO_X$          | SO <sub>X</sub> | PM    | PM <sub>10</sub> | PM <sub>2.5</sub> |
|---------------------------------------|--------|-------|--------|-----------------|-----------------|-------|------------------|-------------------|
| Fuel Combustion                       | 3.84   | 0.55  | 7.87   | 10.08           | 2.62            | 0.26  | 0.26             | 0.26              |
| Waste Disposal                        | 47.10  | 0.97  | 0.48   | 0.30            | 0.05            |       |                  |                   |
| Cleaning and Surface Coatings         | 12.27  | 6.95  | 0.00   | 0.01            |                 |       |                  |                   |
| Petroleum Production and<br>Marketing | 14.01  | 2.34  |        | -               |                 |       |                  | -                 |
| Industrial Processes                  | 2.18   | 1.70  | 0.36   | 0.94            | 0.04            | 1.90  | 1.02             | 0.32              |
| Total Stationary Sources              | 79.40  | 12.51 | 8.72   | 11.33           | 2.71            | 2.16  | 1.28             | 0.58              |
| AREAWIDE SOURCES                      | TOG    | ROG   | со     | NO <sub>X</sub> | SO <sub>X</sub> | PM    | PM <sub>10</sub> | PM <sub>2.5</sub> |
| Solvent Evaporation                   | 17.75  | 15.06 |        |                 |                 |       |                  |                   |
| Miscellaneous Processes               | 9.27   | 2.72  | 27.16  | 3.70            | 0.11            | 37.75 | 20.75            | 6.87              |
| Total Areawide Sources                | 27.03  | 17.79 | 27.16  | 3.70            | 0.11            | 37.75 | 20.75            | 6.87              |
| MOBILE SOURCES                        | TOG    | ROG   | со     | NO <sub>X</sub> | SO <sub>X</sub> | PM    | PM <sub>10</sub> | PM <sub>2.5</sub> |
| On-road Motor Vehicles                | 29.01  | 27.12 | 183.11 | 36.76           | 0.22            | 3.08  | 3.04             | 1.58              |
| Other Mobile Sources                  | 10.70  | 9.64  | 102.38 | 16.06           | 0.10            | 0.89  | 0.87             | 0.81              |
| <b>Total Mobile Sources</b>           | 39.71  | 36.75 | 285.48 | 52.81           | 0.32            | 3.97  | 3.91             | 2.39              |
| TOTAL FOR SANTA CLARA COUNTY          | 146.14 | 67.05 | 321.36 | 67.85           | 3.14            | 43.88 | 25.94            | 9.83              |

Source: California Air Resources Board Almanac Emission Projection Data 2012 Estimated Annual Average Emissions http://www.arb.ca.gov/app/emsinv/emseic1\_query.php (accessed May 2016)

# EXISTING AMBIENT AIR QUALITY: TOXIC AIR CONTAMINANTS

In addition to the criteria air pollutants listed above, another group of pollutants, commonly referred to as toxic air contaminants (TACs) or hazardous air pollutants can result in health effects that can be quite severe. Many TACs are confirmed or suspected carcinogens, or are known or suspected to cause birth defects or neurological damage. Secondly, many TACs can be toxic at very low concentrations. For some chemicals, such as carcinogens, there are no thresholds below which exposure can be considered risk-free.

It is important to understand that TACs are not considered criteria air pollutants and thus are not specifically addressed through the setting of ambient air quality standards. Instead, the EPA and CARB regulate hazardous air pollutants (HAPs) and TACs, respectively, through statutes and regulations that generally require the use of the maximum or best available control technology (MACT and BACT) to limit emissions. These in conjunction with additional rules set forth by the BAAQMD establish the regulatory framework for TACs.

Industrial facilities and mobile sources are significant sources of TACs. The electronics industry, including semiconductor manufacturing, has the potential to contaminate both air and water due to the highly toxic chlorinated solvents commonly used in semiconductor production processes. Sources of TACs go beyond industry. Various common urban facilities also produce TAC emissions, such as gasoline stations

## 5.0 Conservation

(benzene), hospitals (ethylene oxide), and dry cleaners (perchloroethylene). Automobile exhaust also contains TACs such as benzene and 1,3-butadiene. Diesel particulate matter has also been identified as a TAC by the CARB. Diesel PM differs from other TACs in that it is not a single substance but rather a complex mixture of hundreds of substances. BAAQMD research indicates that mobile-source emissions of diesel PM, benzene, and 1,3-butadiene represent a substantial portion of the ambient background risk from TACs in the SFBAAB.

## EXISTING AMBIENT AIR QUALITY: ODORS AND DUST

Other air quality issues of concern in the SFBAAB include nuisance impacts of odors and dust. Objectionable odors may be associated with a variety of pollutants. Common sources of odors include wastewater treatment plants, landfills, composting facilities, refineries and chemical plants. Similarly, nuisance dust may be generated by a variety of sources including quarries, agriculture, grading and construction. Odors rarely have direct health impacts, but they can be very unpleasant and can lead to anger and concern over possible health effects among the public. Each year the BAAQMD receives thousands of citizen complaints about objectionable odors. Dust emissions can contribute to increased ambient concentrations of PM<sub>10</sub>, and can also contribute to reduced visibility and soiling of exposed surfaces.

The Newby Island Landfill is an existing source of odors for the City of Milpitas. The landfill, located at 1601 Dixon Landing Road in the City of San Jose, accepts municipal solid waste, construction/demolition waste, industrial waste, sludge, tires, green materials, and contaminated soils.

All odor complaints by City residents are submitted to the San Jose Local Enforcement Agency. The Bay Area Air Quality Management District investigates odor complaints for the City of Milpitas. The Bay Area Air Quality Management District records from 2013 to 2015 show that 90 percent of confirmed complaints were from the Newby Island Landfill. The Zerowaste Energy Development Facility, an anaerobic digestion facility in San Jose, and the Santa Clara/San Jose Water Pollution Control Plant (WPCP) account for the remaining 10 percent of confirmed odor complaints.

## REFERENCES

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## 5.4 GREENHOUSE GASES AND CLIMATE CHANGE

The Milpitas General Plan will analyze climate change in two ways: 1) The emission of greenhouse gas (GHG) emissions from city and community operations; and 2) Adaptation and resilience to the effects of a changing climate. The emissions of GHGs are addressed in the following section. Impacts related to changes in the climate, such as adaptation and resiliency, are addressed in Section 4.0, Hazards, Safety, and Noise, of this Existing Conditions Report.

### **Greenhouse Gases and Climate Change Linkages**

Various gases in the Earth's atmosphere, classified as atmospheric GHGs, play a critical role in determining the Earth's surface temperature. Solar radiation enters Earth's atmosphere from space, and a portion of the radiation is absorbed by the Earth's surface. The Earth emits this radiation back toward space, but the properties of the radiation change from high-frequency solar radiation to lower-frequency infrared radiation.

Greenhouse gases, which are transparent to solar radiation, are effective in absorbing infrared radiation. As a result, this radiation that otherwise would have escaped back into space is now retained, resulting in a warming of the atmosphere. This phenomenon is known as the greenhouse effect. Among the prominent GHGs contributing to the greenhouse effect are carbon dioxide ( $CO_2$ ), methane ( $CH_4$ ), ozone ( $O_3$ ), water vapor, nitrous oxide ( $O_2$ ), and chlorofluorocarbons ( $CFC_3$ ).

Human-caused emissions of these GHGs, in excess of natural ambient concentrations, are responsible for enhancing the greenhouse effect (Ahrens 2003). Emissions of GHGs contributing to global climate change are attributable in large part to human activities associated with the industrial/manufacturing, utility, transportation, residential, and agricultural sectors (California Energy Commission 2006a). In California, the transportation sector is the largest emitter of GHGs, followed by electricity generation (California Energy Commission 2006a).

As the name implies, global climate change is a global problem. GHGs are global pollutants, unlike criteria air pollutants and toxic air contaminants, which are pollutants of regional and local concern, respectively. California is the 12th to 16th largest emitter of CO<sub>2</sub>in the world.

Carbon dioxide equivalents are a measurement used to account for the fact that different GHGs have different potential to retain infrared radiation in the atmosphere and contribute to the greenhouse effect. This potential, known as the global warming potential of a GHG, is also dependent on the lifetime, or persistence, of the gas molecule in the atmosphere. Expressing GHG emissions in carbon dioxide equivalents takes the contribution of all GHG emissions to the greenhouse effect and converts them to a single unit equivalent to the effect that would occur if only CO<sub>2</sub> were being emitted.

## **Effects of Global Climate Change**

The effects of increasing global temperature are far reaching and extremely difficult to quantify. The scientific community continues to study the effects of global climate change. In general, increases in the ambient global temperature as a result of increased GHGs is anticipated to result in rising sea levels, which could threaten coastal areas through accelerated coastal erosion, threats to levees and inland water systems, and disruption to coastal wetlands and habitat.

If the temperature of the ocean warms, it is anticipated that the winter snow season would be shortened. Snowpack in the Sierra Nevada provides both water supply (runoff) and storage (within the

snowpack before melting), which is a major source of supply for the State. According to a California Energy Commission report, the snowpack portion of the supply could potentially decline by 70 to 90 percent by the end of the 21<sup>st</sup> century (CEC 2006c). This phenomenon could lead to significant challenges securing an adequate water supply for a growing State population. Further, the increased ocean temperature could result in increased moisture flux into the State; however, since this would likely increasingly come in the form of rain rather than snow in the high elevations, increased precipitation could lead to increased potential and severity of flood events, placing more pressure on California's levee/flood control system.

Sea level has risen approximately seven inches during the last century and, according to the CEC report, it is predicted to rise an additional 22 to 35 inches by 2100, depending on the future GHG emissions levels (CEC 2006c). If this occurs, resultant effects could include increased coastal flooding, saltwater intrusion, and disruption of wetlands (CEC 2006c). As the existing climate throughout California changes over time, mass migration of species, or failure of species to migrate in time to adapt to the perturbations in climate, could also result. Under the emissions scenarios of the Climate Scenarios report (California Climate Change Center 2006), the impacts of global warming in California are anticipated to include, but are not limited to, the following.

### PUBLIC HEALTH

Higher temperatures are expected to increase the frequency, duration, and intensity of conditions conducive to air pollution formation. For example, days with weather conducive to ozone formation are projected to increase from 25 to 35 percent under the lower warming range, to 75 to 85 percent under the medium warming range. In addition, if global background ozone levels increase as predicted in some scenarios, it may become impossible to meet local air quality standards. Air quality could be further compromised by increases in wildfires, which emit fine particulate matter that can travel long distances depending on wind conditions. The Climate Scenarios report indicates that large wildfires could become up to 55 percent more frequent if GHG emissions are not significantly reduced.

In addition, under the higher warming scenario, there could be up to 100 more days per year with temperatures above 90°F in Los Angeles and 95°F in Sacramento by 2100. This is a large increase over historical patterns and approximately twice the increase projected if temperatures remain within or below the lower warming range. Rising temperatures will increase the risk of death from dehydration, heat stroke/exhaustion, heart attack, stroke, and respiratory distress caused by extreme heat.

### WATER RESOURCES

A vast network of man-made reservoirs and aqueducts capture and transport water throughout the State from northern California rivers and the Colorado River. The current distribution system relies on Sierra Nevada snow pack to supply water during the dry spring and summer months. Rising temperatures, potentially compounded by decreases in precipitation, could severely reduce spring snow pack, increasing the risk of summer water shortages.

The State's water supplies are also at risk from rising sea levels. An influx of saltwater would degrade California's estuaries, wetlands, and groundwater aquifers. Saltwater intrusion caused by rising sea levels is a major threat to the quality and reliability of water within the southern edge of the Sacramento/San Joaquin River Delta, a major state fresh water supply. Global warming is also projected to seriously affect agricultural areas, with California farmers projected to lose as much as 25 percent of the water supply they need; decrease the potential for hydropower production within the state

## 5.0 Conservation

(although the effects on hydropower are uncertain); and seriously harm winter tourism. Under the lower warming range, the ski season at lower elevations could be reduced by as much as 1 month. If temperatures reach the higher warming range and precipitation declines, there might be many years with insufficient snow for skiing and snowboarding.

If GHG emissions continue unabated, more precipitation will fall as rain instead of snow, and the snow that does fall will melt earlier, reducing the Sierra Nevada spring snow pack by as much as 70 to 90 percent. Under the lower warming scenario, snow pack losses are expected to be only half as large as those expected if temperatures were to rise to the higher warming range. How much snow pack will be lost depends in part on future precipitation patterns, the projections for which remain uncertain. However, even under the wetter climate projections, the loss of snow pack would pose challenges to water managers, hamper hydropower generation, and nearly eliminate all skiing and other snow-related recreational activities.

### AGRICULTURE

Increased GHG emissions are expected to cause widespread changes to the agriculture industry, reducing the quantity and quality of agricultural products statewide. Although higher carbon dioxide levels can stimulate plant production and increase plant water-use efficiency, California's farmers will face greater water demand for crops and a less reliable water supply as temperatures rise. Crop growth and development will change, as will the intensity and frequency of pest and disease outbreaks. Rising temperatures will likely aggravate ozone pollution, which makes plants more susceptible to disease and pests and interferes with plant growth.

Plant growth tends to be slow at low temperatures, increasing with rising temperatures up to a threshold. However, faster growth can result in less-than optimal development for many crops, so rising temperatures are likely to worsen the quantity and quality of yield for a number of California's agricultural products. Products likely to be most affected include wine grapes, fruits and nuts, and milk.

In addition, continued global warming will likely shift the ranges of existing invasive plants and weeds and alter competition patterns with native plants. Range expansion is expected in many species while range contractions are less likely in rapidly evolving species with significant populations already established. Should range contractions occur, it is likely that new or different weed species will fill the emerging gaps. Continued global warming is also likely to alter the abundance and types of many pests, lengthen pests' breeding seasons, and increase pathogen growth rates.

### FORESTS AND LANDSCAPES

Global warming is expected to intensify this threat by increasing the risk of wildfire and altering the distribution and character of natural vegetation. If temperatures rise into the medium warming range, the risk of large wildfires in California could increase by as much as 55 percent, which is almost twice the increase expected if temperatures stay in the lower warming range. However, since wildfire risk is determined by a combination of factors, including precipitation, winds, temperature, and landscape and vegetation conditions, future risks will not be uniform throughout the state. For example, if precipitation increases as temperatures rise, wildfires in southern California are expected to increase by approximately 30 percent toward the end of the century. In contrast, precipitation decreases could increase wildfires in northern California by up to 90 percent.

Moreover, continued global warming will alter natural ecosystems and biological diversity within the state. For example, alpine and sub-alpine ecosystems are expected to decline by as much as 60 to 80

percent by the end of the century as a result of increasing temperatures. The productivity of the state's forests is also expected to decrease as a result of global warming.

RISING SEA LEVELS

Rising sea levels, more intense coastal storms, and warmer water temperatures will increasingly threaten the state's coastal regions. Under the higher warming scenario, sea level is anticipated to rise 22 to 35 inches by 2100. Elevations of this magnitude would inundate coastal areas with saltwater, accelerate coastal erosion, threaten vital levees and inland water systems, and disrupt wetlands and natural habitats.

### REGULATORY SETTING

**FEDERAL** 

### **Federal Climate Change Policy**

According to the EPA, "the United States government has established a comprehensive policy to address climate change" that includes slowing the growth of emissions; strengthening science, technology, and institutions; and enhancing international cooperation. To implement this policy, "the Federal government is using voluntary and incentive-based programs to reduce emissions and has established programs to promote climate technology and science." The Federal government's goal is to reduce the greenhouse gas (GHG) intensity (a measurement of GHG emissions per unit of economic activity) of the American economy by 18 percent over the 10-year period from 2002 to 2012. In addition, the EPA administers multiple programs that encourage voluntary GHG reductions, including "ENERGY STAR," "Climate Leaders," and Methane Voluntary Programs. However, as of this writing, there are no adopted Federal plans, policies, regulations, or laws directly regulating GHG emissions.

## **Energy Policy Act of 2005**

The Energy Policy Act of 2005 was signed into law on August 8, 2005. Generally, the act provides for renewed and expanded tax credits for electricity generated by qualified energy sources, such as landfill gas; provides bond financing, tax incentives, grants, and loan guarantees for a clean renewable energy and rural community electrification; and establishes a Federal purchase requirement for renewable energy.

STATE

### **Assembly Bill 1493**

In response to AB 1493, CARB approved amendments to the California Code of Regulations (CCR) adding GHG emission standards to California's existing motor vehicle emission standards. Amendments to CCR Title 13 Sections 1900 (CCR 13 1900) and 1961 (CCR 13 1961), and adoption of Section 1961.1 (CCR 13 1961.1), require automobile manufacturers to meet fleet average GHG emission limits for all passenger cars, light-duty trucks within various weight criteria, and medium-duty passenger vehicle weight classes beginning with the 2009 model year. Emission limits are further reduced each model year through 2016. For passenger cars and light-duty trucks 3,750 pounds or less loaded vehicle weight (LVW), the 2016 GHG emission limits are approximately 37 percent lower than during the first year of the regulations in 2009. For medium-duty passenger vehicles and light-duty trucks 3,751 LVW to 8,500 pounds gross vehicle weight (GVW), GHG emissions are reduced approximately 24 percent between 2009 and 2016.

CARB requested a waiver of Federal preemption of California's Greenhouse Gas Emissions Standards. The intent of the waiver is to allow California to enact emissions standards to reduce carbon dioxide and other greenhouse gas emissions from automobiles in accordance with the regulation amendments to the CCRs that fulfill the requirements of AB 1493. The EPA granted a waiver to California to implement its greenhouse gas emissions standards for cars.

## California Executive Orders S-3-05 and S-20-06, and Assembly Bill 32

On June 1, 2005, Governor Arnold Schwarzenegger signed Executive Order S-3-05. The goal of this Executive Order is to reduce California's GHG emissions to: 1) 2000 levels by 2010, 2) 1990 levels by 2020 and 3) 80% below 1990 levels by 2050.

In 2006, this goal was further reinforced with the passage of Assembly Bill 32 (AB 32), the Global Warming Solutions Act of 2006. AB 32 sets the same overall GHG emissions reduction goals while further mandating that CARB create a plan, which includes market mechanisms, and implement rules to achieve "real, quantifiable, cost-effective reductions of greenhouse gases." Executive Order S-20-06 further directs State agencies to begin implementing AB 32, including the recommendations made by the State's Climate Action Team.

## **Assembly Bill 1007**

Assembly Bill 1007 (Pavley, Chapter 371, Statutes of 2005) directed the CEC to prepare a plan to increase the use of alternative fuels in California. As a result, the CEC prepared the State Alternative Fuels Plan in consultation with State, Federal, and local agencies. The plan presents strategies and actions California must take to increase the use of alternative non-petroleum fuels in a manner that minimizes costs to California and maximizes the economic benefits of in-state production. The Plan assessed various alternative fuels and developed fuel portfolios to meet California's goals to reduce petroleum consumption, increase alternative fuels use, reduce greenhouse gas emissions, and increase in-state production of biofuels without causing a significant degradation of public health and environmental quality.

## **Governor's Low Carbon Fuel Standard (Executive Order S-01-07)**

Executive Order S-01-07 establishes a statewide goal to reduce the carbon intensity of California's transportation fuels by at least 10 percent by 2020 through establishment of a Low Carbon Fuel Standard. The Low Carbon Fuel Standard is incorporated into the State Alternative Fuels Plan and is one of the proposed discrete early action GHG reduction measures identified by CARB pursuant to AB 32.

### **Climate Action Program at Caltrans**

Caltrans prepared a Climate Action Program in response to new regulatory directives. The goal of the Climate Action Program is to promote clean and energy efficient transportation, and provide guidance for mainstreaming energy and climate change issues into business operations. The overall approach to lower fuel consumption and  $CO_2$  from transportation is twofold: (1) reduce congestion and improve efficiency of transportation systems through smart land use, operational improvements, and Intelligent Transportation Systems; and (2) institutionalize energy efficiency and GHG emission reduction measures and technology into planning, project development, operations, and maintenance of transportation facilities, fleets, buildings, and equipment.

The reasoning underlying the Climate Action Program is the conclusion that "the most effective approach to addressing GHG reduction, in the short-to-medium term, is strong technology policy and market mechanisms to encourage innovations. Rapid development and availability of alternative fuels

and vehicles, increased efficiency in new cars and trucks (light and heavy duty), and super clean fuels are the most direct approach to reducing GHG emissions from motor vehicles (emission performance standards and fuel or carbon performance standards)."

### Senate Bill 97

Senate Bill 97 (Chapter 185, 2007) required the Governor's Office of Planning and Research (OPR) to develop recommended amendments to the State CEQA Guidelines for addressing greenhouse gas emissions. OPR prepared its recommended amendments to the State CEQA Guidelines to provide guidance to public agencies regarding the analysis and mitigation of greenhouse gas emissions and the effects of greenhouse gas emissions in draft CEQA documents. The Amendments became effective on March 18, 2010.

### Senate Bill 375

SB 375 requires CARB to develop regional greenhouse gas emission reduction targets to be achieved from the automobile and light truck sectors for 2020 and 2035. The 18 metropolitan planning organizations (MPO) in California will prepare a "sustainable communities strategy" to reduce the amount of greenhouse gas emission in their respective regions and demonstrate the ability for the region to attain CARB's reduction targets. CARB would later determine if each region is on track to meet their reduction targets. In addition, cities would have extra time -- eight years instead of five -- to update housing plans required by the State.

## EXISTING GREENHOUSE GAS EMISSIONS IN MILPITAS

The City adopted a Climate Action Plan (CAP) in May 2013. The CAP is designed to streamline environmental review of future development projects in the City of Milpitas, consistent with CEQA Guidelines Section 15183.5(b). The CAP identifies a strategy, baseline emissions and forecasts, reduction measures, and implementation strategies the City is currently using to achieve the State-recommended greenhouse gas (GHG) emissions reduction target of 15% below 2005 emission levels by 2020.

### **Baseline Emissions**

In 2005, the Milpitas community emitted approximately 744,150 metric tons of  $CO_2$  equivalent (MT  $CO_2$ e). Table 5.4-1 reports these emissions by sector and ranks the sectors from highest to lowest.

TABLE 5.4-1: BASELINE GREENHOUSE GAS EMISSIONS BY SECTOR

| EMISSIONS SECTOR       | 2005 MT CO <sub>2</sub> E | PERCENTAGE OF TOTAL |
|------------------------|---------------------------|---------------------|
| Transportation         | 320,990                   | 43%                 |
| Non-residential Energy | 183,800                   | 25%                 |
| Stationary Sources     | 101,480                   | 14%                 |
| Residential Energy     | 64,230                    | 9%                  |
| Solid Waste            | 54,410                    | 7%                  |
| Off-Road Equipment     | 15,140                    | 2%                  |
| Water and Wastewater   | 2,410                     | <1%                 |
| Light Rail             | 1,070                     | <1%                 |
| Direct Wastewater      | 620                       | <1%                 |
| Total*                 | 744,150                   | 100%                |

Note: \*Due to rounding, the total may not equal the sum of component parts

Source: City of Milpitas Climate Action Plan (2013).

Table 5.4-1 reports stationary source emissions, which include those from the Newby Island Resource Recovery Park, and direct wastewater emissions. Stationary sources are fixed emitters of air pollutants, such as power plants, stationary generators, petrochemical plants, and other heavy industrial sources. Since stationary source emissions are influence by market forces beyond the City's local influence and are best regulated by the BAAQMD or through federal and state programs, they are reported in this Inventory for informational purposes only. Similarly, the City has limited control over the operation of the Newby Island Resource Recovery Park and is unable to directly affect the emissions generated from previously generated waste. The baseline inventory is intended to guide future local policy decisions that relate to emissions within the City's influence; therefore, stationary sources, and direct landfill emissions are excluded from all further discussions in this Inventory.

Table 5.4-2 reflect Milpitas' jurisdictional baseline of 642,670 MT  $CO_2e$ . Transportation was the largest sector (320,990 MT  $CO_2e$ ), contributing about 50% of total emissions. Energy use was the second largest sector (248,030 MT  $CO_2e$ , 39%). Of these emissions, nonresidential energy use (183,800 MT  $CO_2e$ , 29%) comprised a greater percentage than residential energy use (64,230 MT  $CO_2e$ , 10%). The remaining 11% of emissions came from solid waste (54,410 MT  $CO_2e$ , 8%), water and wastewater (2,410 MT  $CO_2e$ , less than 1%), light rail (1,070 MT  $CO_2e$ , less than 1%), and direct wastewater emissions (620 MT  $CO_2e$ ).

TABLE 5.4-2: JURISDICTIONAL BASELINE GREENHOUSE GAS EMISSIONS BY SECTOR

| EMISSIONS SECTOR       | 2005 MT CO₂E | PERCENTAGE OF TOTAL |
|------------------------|--------------|---------------------|
| Transportation         | 320,990      | 50%                 |
| Non-residential Energy | 183,800      | 29%                 |
| Residential Energy     | 64,230       | 10%                 |
| Solid Waste            | 54,410       | 8%                  |
| Off-Road Equipment     | 15,140       | 2%                  |
| Water and Wastewater   | 2,410        | <1%                 |
| Light Rail             | 1,070        | <1%                 |
| Direct Wastewater      | 620          | <1%                 |
| Total*                 | 642,670      | 100%                |

Note: \*Due to rounding, the total may not equal the sum of component parts

Source: City of Milpitas Climate Action Plan (2013).

### **Reduction Target**

The Milpitas CAP established a local GHG reduction target of 15% below baseline 2005 emissions levels by 2020. The AB32 Climate Change Scoping Plan provides substantial evidence supporting use of this target by the City of Milpitas. This target serves as the City's cumulative level of significance for community-wide GHG emissions through 2020. The reduction target equates to a 93,940 MT CO₂e reduction in community-wide GHGs from baseline 2005 levels by 2020. It will require a reduction of 79,990 MT CO₂e from 2020 adjusted BAU forecast levels. Table 5.4-3 identifies the 3% reduction from baseline emissions that the Milpitas CAP anticipates with implementation of State of California policies and programs, and the 12% gap that local GHG reduction measures will address to achieve the 15% reduction target.

TABLE 5.4-3: GREENHOUSE GAS EMISSIONS TARGET AND NECESSARY LOCAL REDUCTION

| EMISSIONS SECTOR                       | 2020 MT CO₂E |
|--|--------------|
| Reduction Target (15% below baseline)  | 546,270      |
| Adjusted Business-as-Usual Forecast    | 626,260      |
| Local Reduction Needed to Reach Target | -79,990      |

Source: City of Milpitas Climate Action Plan (2013).

### **GHG Reduction Measures**

The Milpitas CAP provides the following local GHG reduction measures, with the expected quantified GHG Reduction (in MT CO₂e) in parenthesis:

- Energy:
  - Goal 1: Energy Efficiency in Existing Development (25,240)
  - o Goal 2: Energy Efficiency in New Development (150)
  - o Goal 3: Renewable Energy (15,200)
- Water:
  - Goal 4: Water Conservation (<10)</li>
- Transportation & Land Use:
  - Goal 5: Mixed-Use Development (Not Quantified)
  - o Goal 6: Transportation-oriented Development (12,350)
  - o Goal 7: Bicycle- and Pedestrian-oriented Development (Not Quantified)
  - Goal 8: Ridesharing and Transit (4,230)
  - o Goal 9: Parking (Not Quantified)
  - Goal 10: Alternative Fuels and Ridesharing (3,590)
- Solid Waste:
  - Solid Waste Diversion (9,200)
- Off-Road Equipment:
  - o Off-Road Equipment (4,260)

The total local-level reductions provided in the Milpitas CAP would reduce GHG emissions by 87,450 MT CO₂e in 2020, from baseline year 2005.

## **Reviewing Development Projects Under the CAP**

For discretionary projects seeking to use CEQA streamlining provisions, the City may require measures in the Milpitas CAP as mandatory conditions of approval or as mitigation identified in a mitigated negative declaration or in an environmental impact report, as feasible, on a project-by-project basis. This approach allows the City to ensure that new development can benefit from CEQA streamlining provisions while also ensuring that the City can achieve the reduction targets outlined in this plan.

Furthermore, as a programmatic tiering document under CEQA, the CAP is the City's one-stop shop for greenhouse gas analysis and mitigation under CEQA. This CAP does not identify measures as mandatory or voluntary. Rather, the City ensures appropriate use of the CAP for CEQA streamlining by maintaining the prerogative to identify appropriate mandatory and voluntary measures to integrate into project design or mitigation. The City recommends inclusion of all feasible and applicable measures on a project-by-project basis. The City uses the development checklist provided within the CAP to identify appropriate measures. City staff also works with project applicants to determine the appropriate use of the CEQA benefits of the Climate Action Plan.

## 5.0 Conservation

## **Monitoring and Implementation**

The Milpitas CAP identifies the procedures the City uses to monitor implementation of the CAP and presents methods for evaluating the effectiveness of CAP measures, as well as potential reasons to reevaluate reduction measures in the future. These procedures are consistent with State CEQA Guidelines Section 15183.5(b)(I)(E). The CAP also identifies the standards the City will implement on a case-by-case basis and presents initial milestones the City must accomplish to begin using the CAP as a basis for project-level CEQA review.

The CAP requires that City leadership to execute these measures and report on their progress. The CAP identifies the responsible department for each measure and offers time frames for implementing each strategy. Lastly, successful implementation requires regular reporting. Staff monitors progress toward implementing the CAP on an annual basis and reports progress to the City Council each year.

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